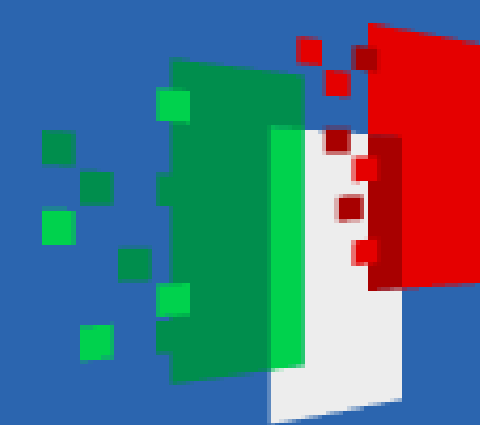




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# Producing Plasma Electrolytic Oxidation (PEO) corrosion resistant coatings on aluminium 2024 texturized with a riblet-like surface for aeronautical applications

**Matteo Gamba**<sup>1</sup>, Marco Ormellese<sup>1</sup>, Andrea Cristoforetti<sup>2</sup>, Michele Fedel<sup>2</sup>, Andrea Brenna<sup>1</sup>

<sup>1</sup> *Dipartimento di Chimica, Materiali ed Ingegneria Chimica “Giulio Natta”, Politecnico di Milano*

<sup>2</sup> *Dipartimento di Ingegneria Industriale, Università di Trento*

**EMCR 2025**

**San Servolo (VE) | 15-19 giugno 2025**



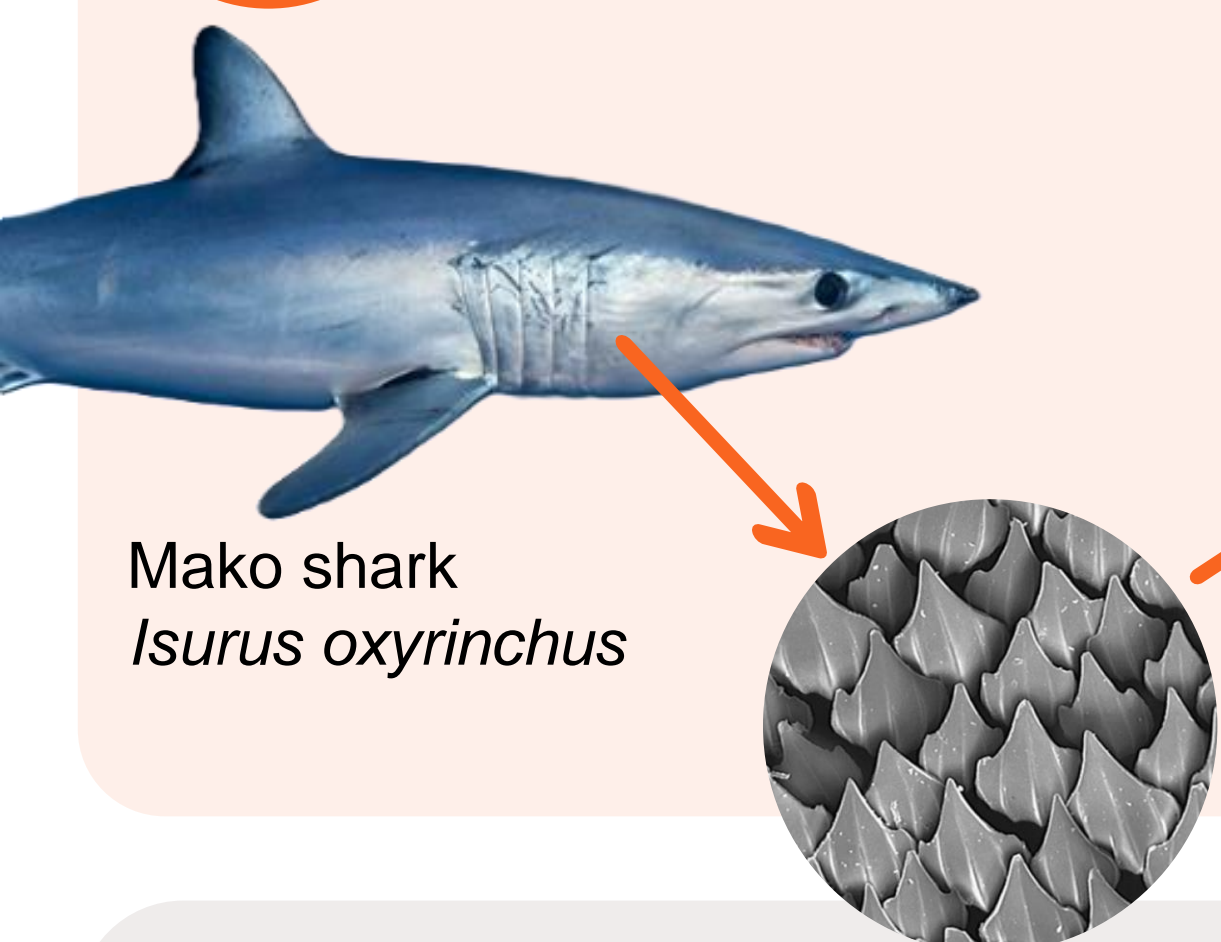
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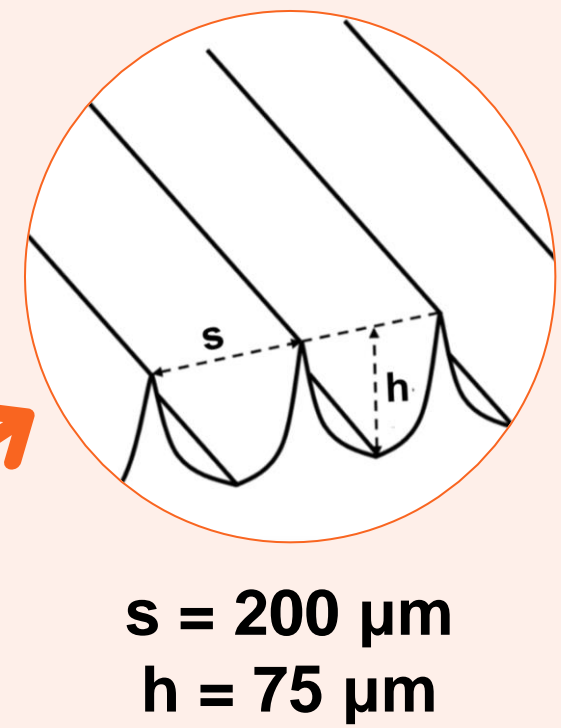
<sup>1</sup> Politecnico di Milano, Dipartimento di Chimica, Materiali e Ingegneria Chimica “Giulio Natta”, Milano

<sup>2</sup> Università di Trento, Dipartimento di Ingegneria Industriale, Trento

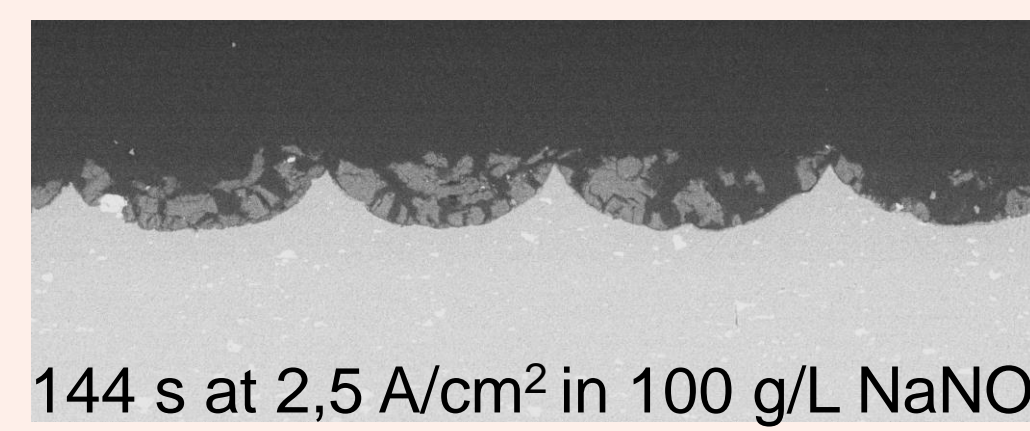
## Project outline



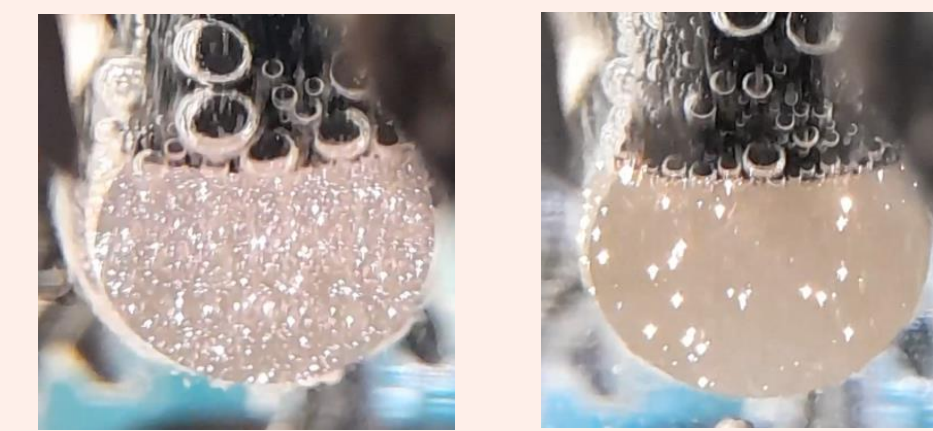
- Sharkskin biomimetic textures are able to reduce fluid dynamic drag during motion.
- Plasma Electrolytic Oxidation (PEO) coatings proved to be effective for corrosion protection on aluminum alloys.
- Their combination would offer desirable surfaces for aeronautical vehicles.



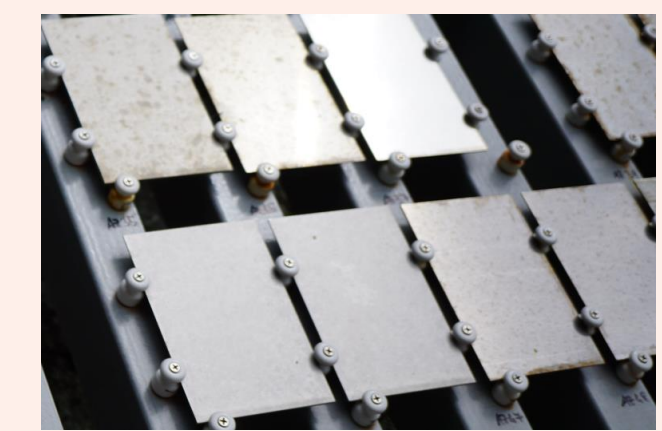
Biomimetic texture production by electrochemical etching



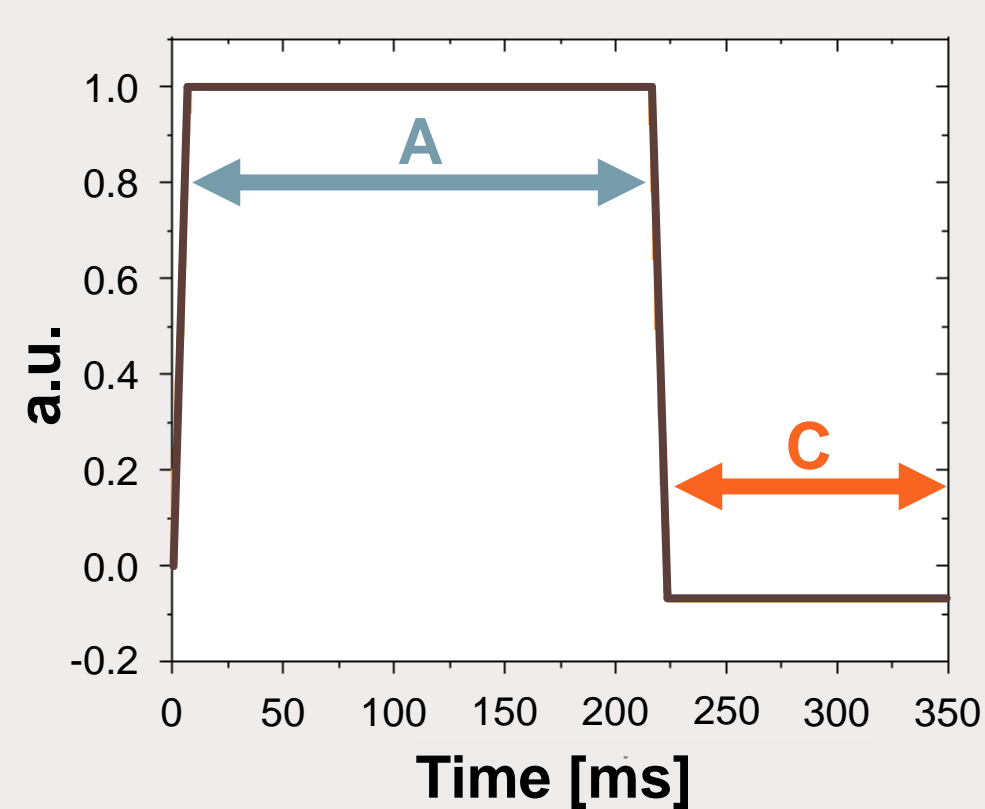
PEO coating optimization for textured surfaces



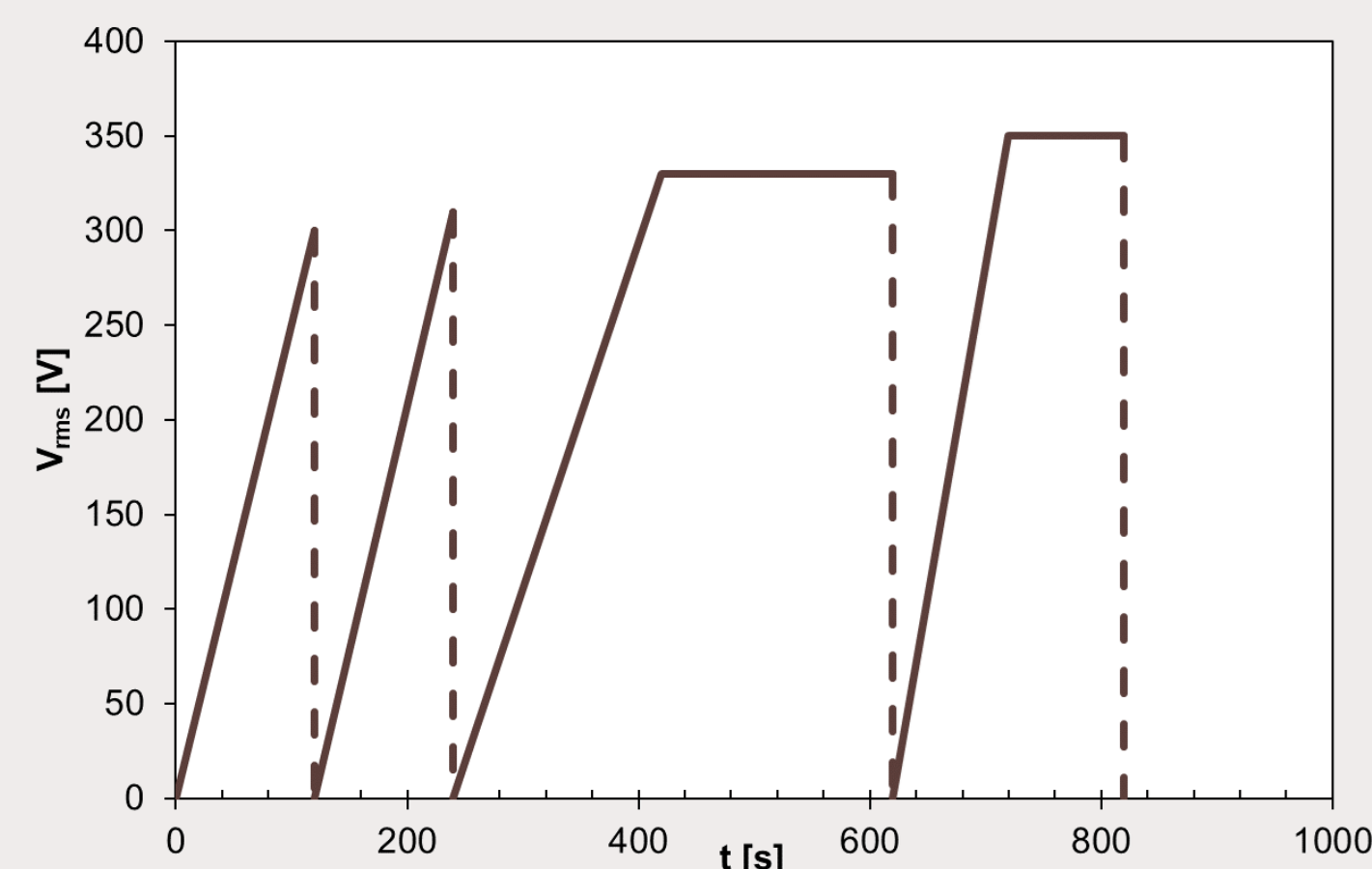
Corrosion testing of surfaces



## Materials Methods



Waveform used for PEO electrical input. Square signal, 60% time in anodic polarization, cathodic amplitude 7% of anodic one.



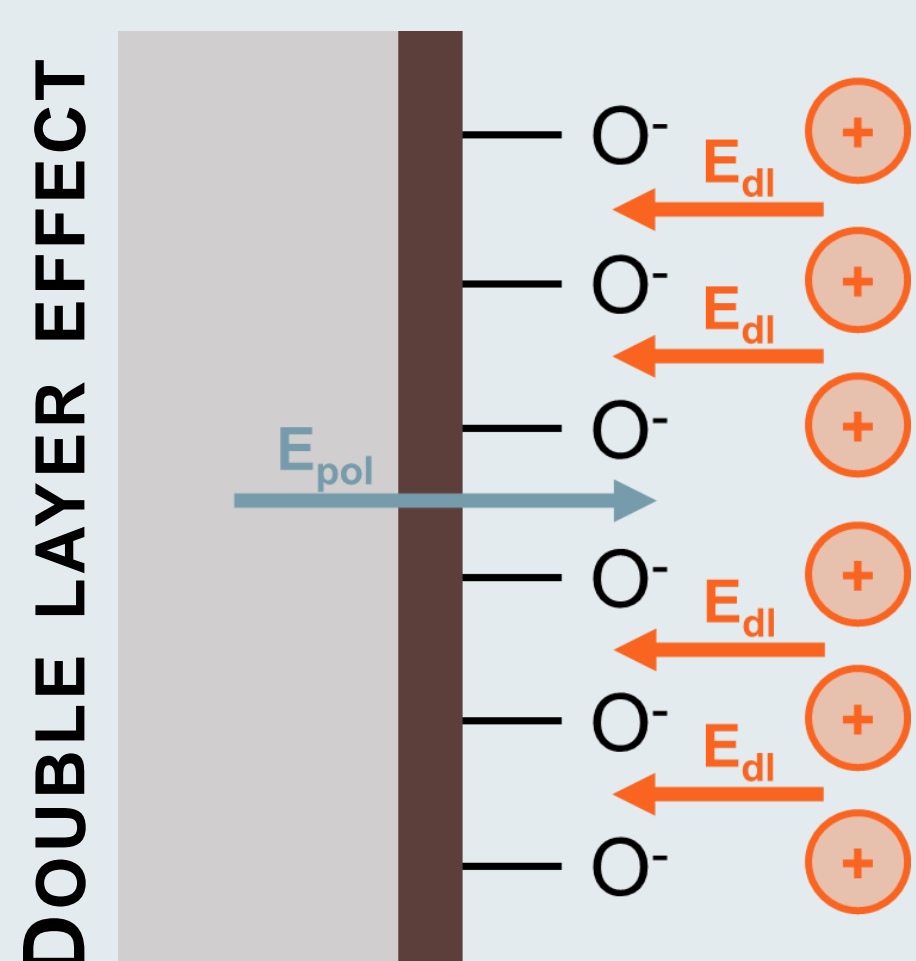
Multistep electrical input used for all the PEO coatings. Four steps encompassing ramps and maintenance periods, with maximum voltage 350 V<sub>rms</sub> and total duration around 15 min.

Electrolyte	Alkali			Additives	
	KOH, mol/L	NaOH, mol/L	CH <sub>3</sub> COOH, g/L	Na <sub>2</sub> SiO <sub>3</sub> , g/L	Glycerin, g/L
0.20 KS	0.20	-	-	10	10
0.20 KNS	0.04	0.16	-	10	10
0.09 KNS	0.014	0.076	-	10	10
0.09 KNSA	0.014	0.076	10	10	10

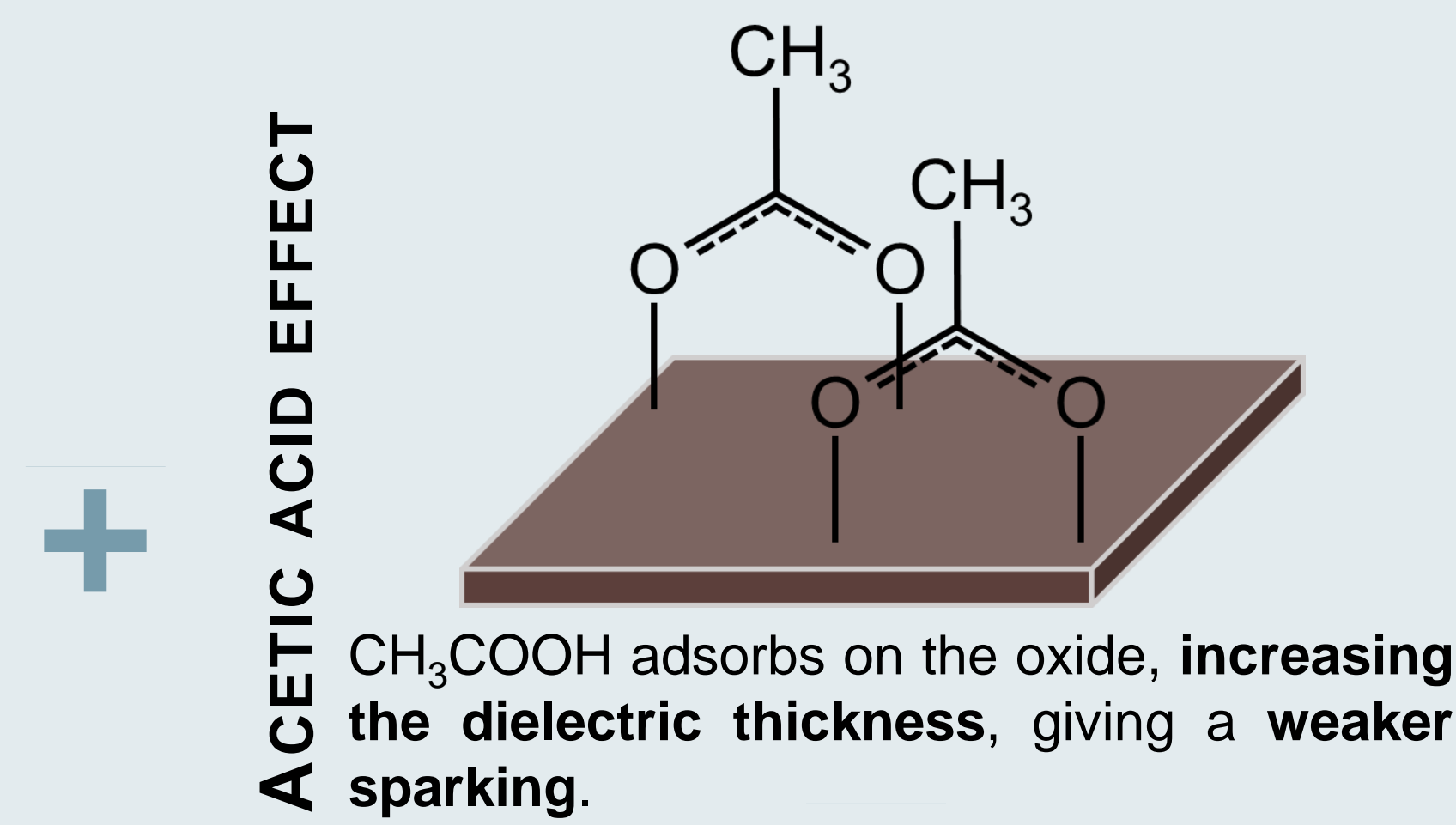
Electrolytic solutions used for PEO coatings.

## Discharge features

- More intense sparking corresponds to more defected, but thicker, coatings.
- Spark intensity depends on oxide thickness, electrolyte resistivity and double layer formation.
- Spark characteristics can be modified by acting on the electrolytic solution composition.



Due to the lower mobility of Na<sup>+</sup> with respect to K<sup>+</sup>, double layer formation is slower in 0.20 KNS, giving a stronger, but more homogeneous, sparking.



ALKALI CONCENTRATION EFFECT  
Reducing alkali concentration the electrolyte becomes resistive, obtaining weaker sparks.

Stronger plasma sparks  
Thicker coating  
Larger defects

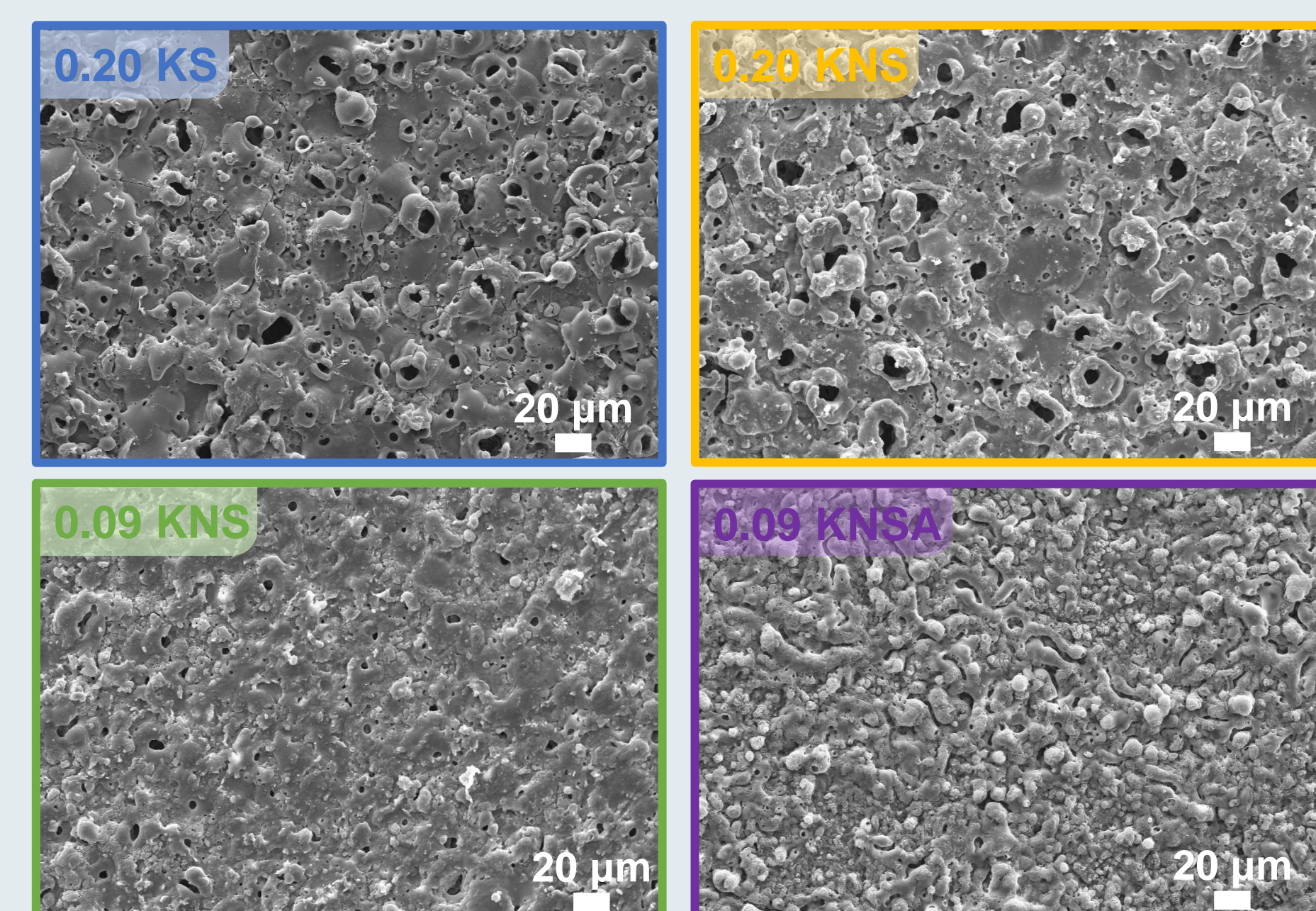
0.20 KS

0.20 KNS

0.09 KNS

0.09 KNSA

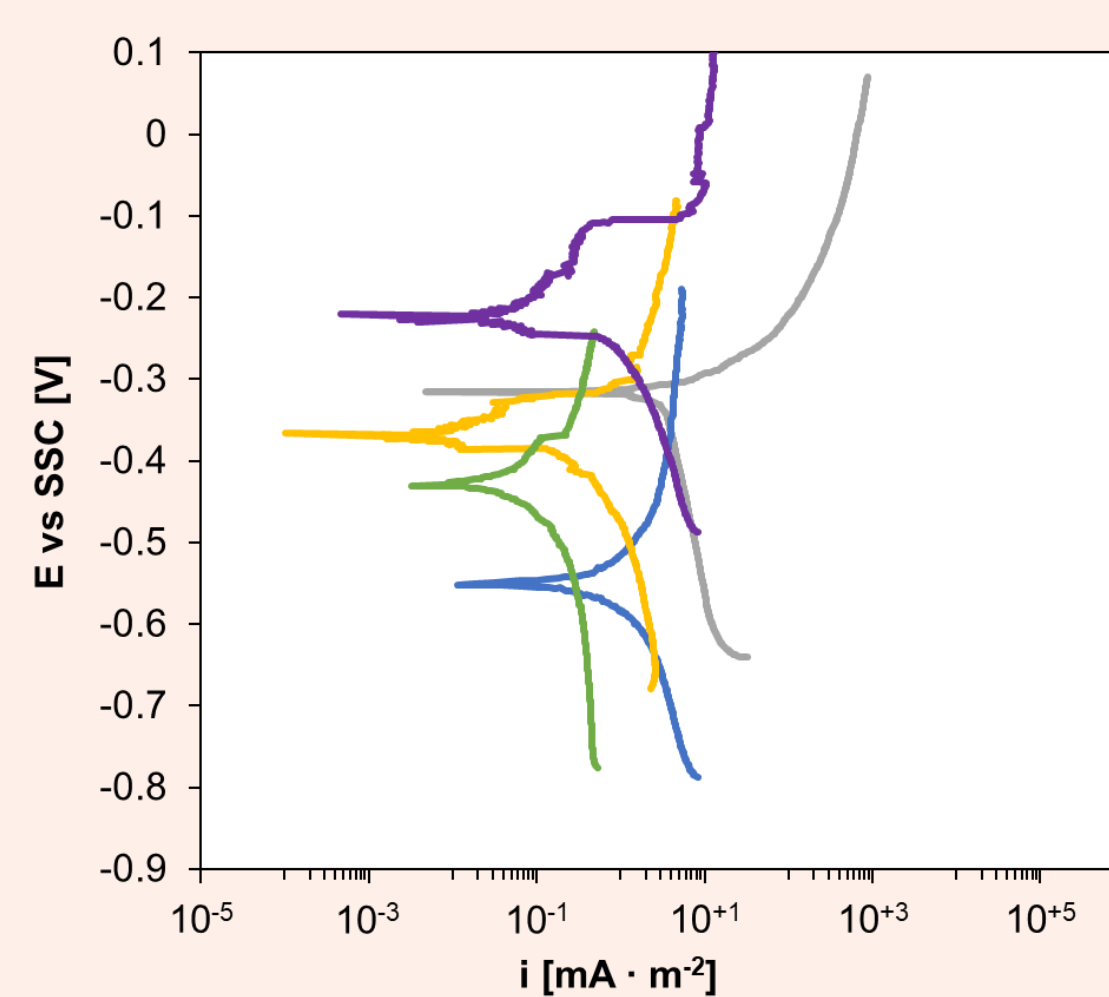
Weaker plasma sparks  
Thinner coating  
Smaller defects



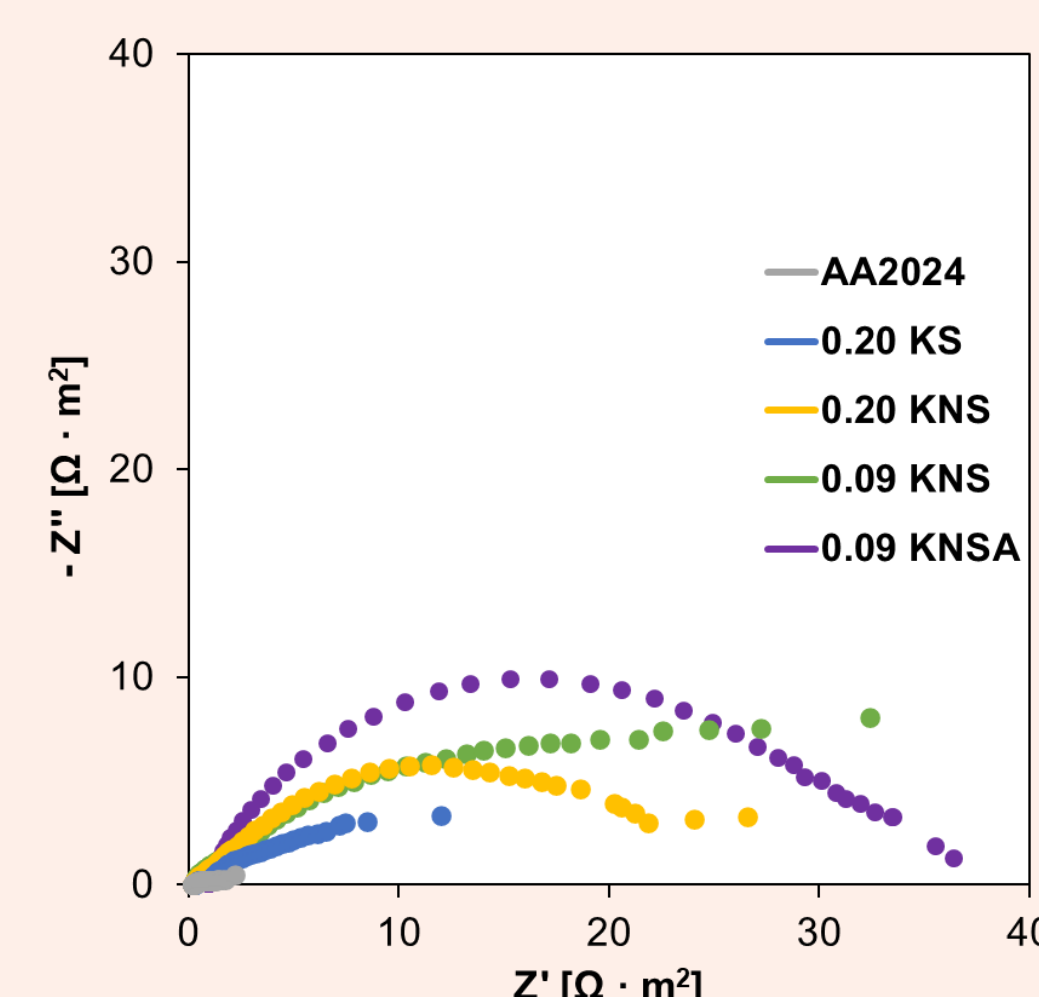
SEM 1000x magnifications of coatings. Pores and cracks amount decreases with spark intensity.

## Corrosion resistance

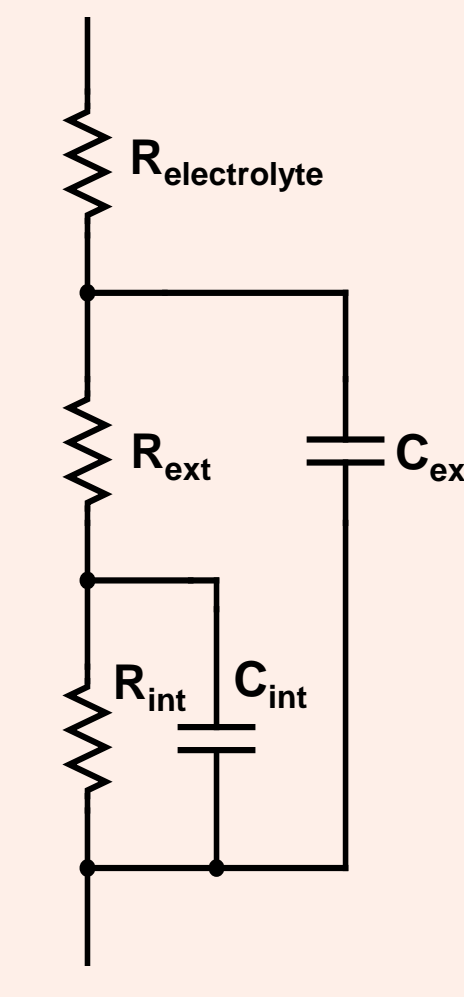
- All the coatings reduce  $i_{corr}$  of AA2024 by at least 10 times.
- Electrochemical testing shows that less defected coatings are more protective, irrespectively of their thickness.



Potentiodynamic polarization  
20 ppm Cl<sup>-</sup>, 1 h immersion



EIS  
20 ppm Cl<sup>-</sup>, 1 h immersion  
10<sup>5</sup> to 10<sup>-2</sup> Hz



All coatings are bilayered

Larger defects  
Higher  $i_{corr}$   
Lower  $R_p$

0.20 KS

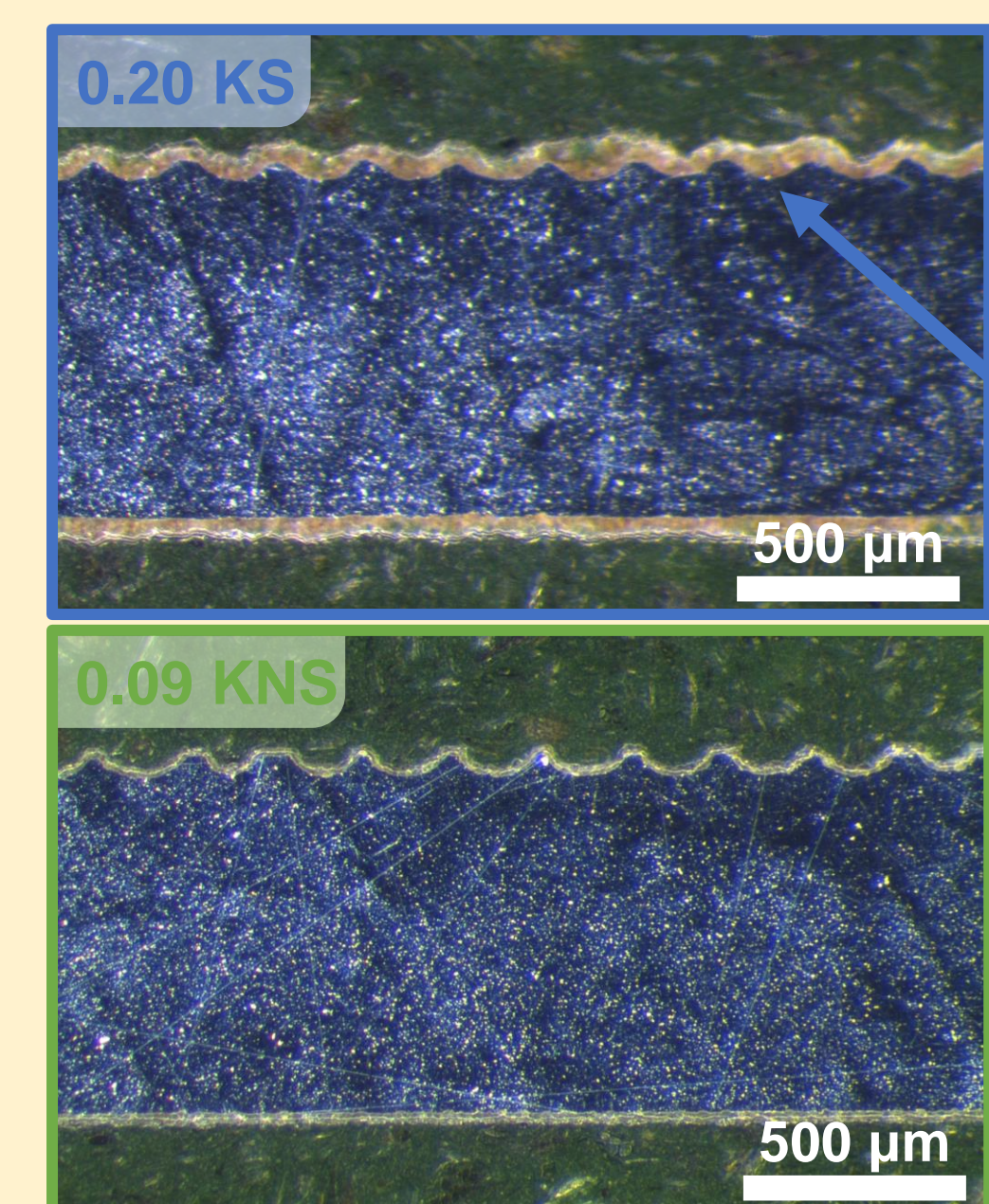
0.20 KNS

0.09 KNS

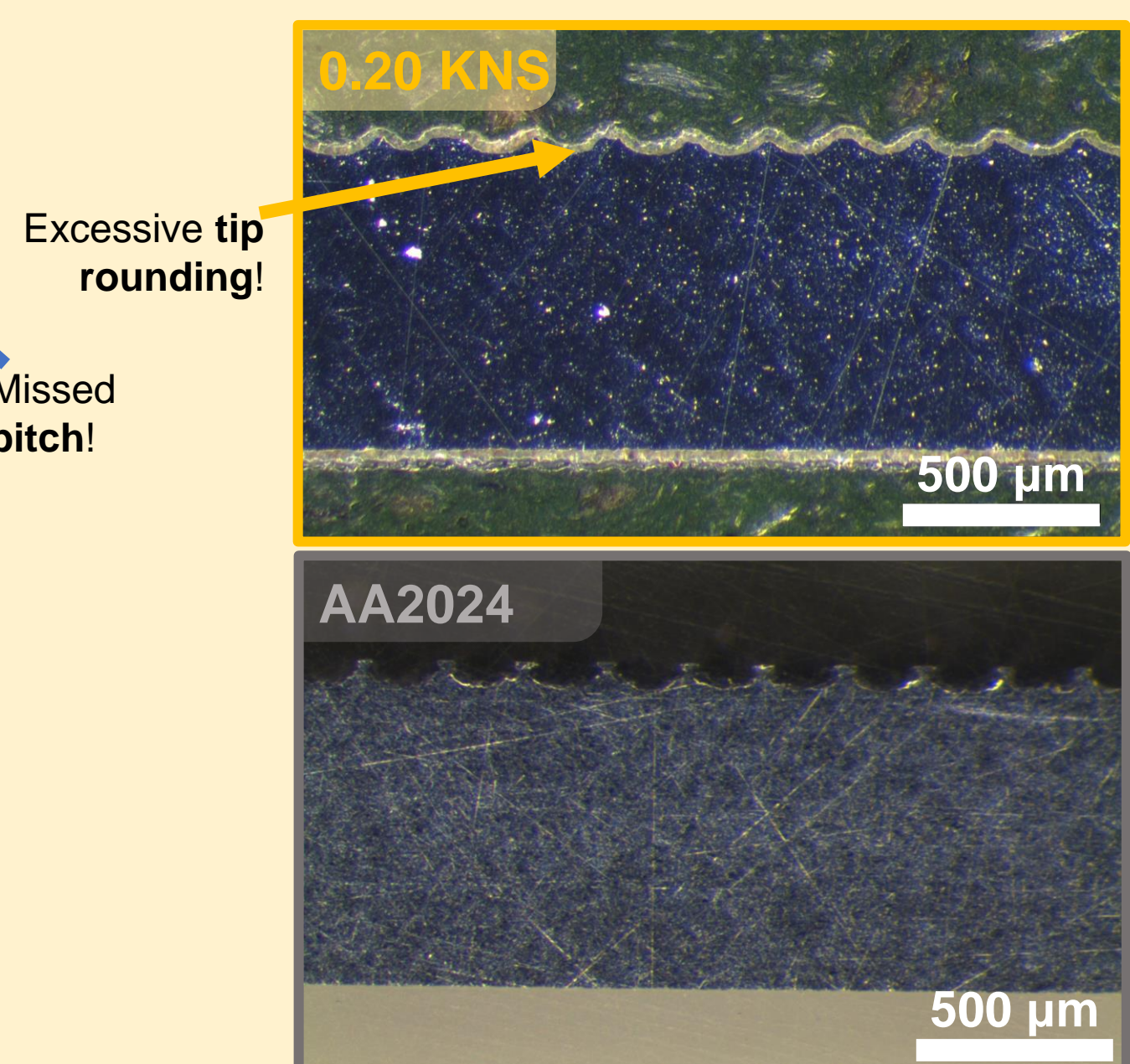
0.09 KNSA

Smaller defects  
Lower  $i_{corr}$   
Higher  $R_p$

## Sharkskin Texture



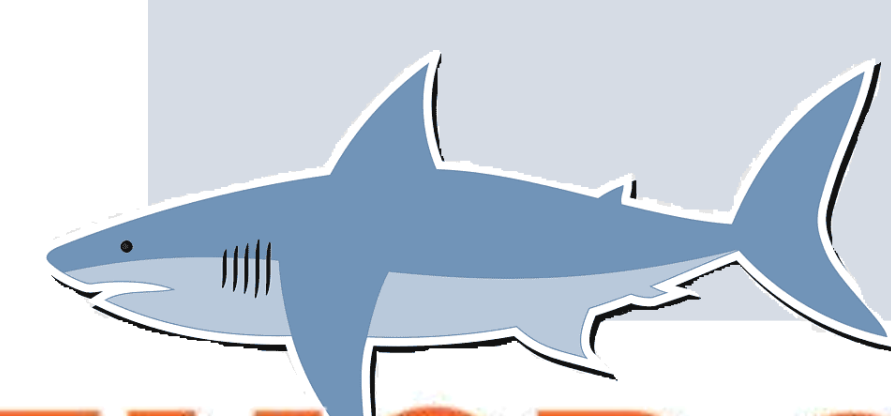
- Not all the processes are accurate.
- Electrolytes able at reducing spark intensity give better reproduction.



## Conclusions and future development

- The amount of coating defects decreases by **decreasing spark intensity** during PEO. Thus, coating homogeneity, **corrosion resistance** and **texture reproduction** are improved.
- Substituting K<sup>+</sup> cations with Na<sup>+</sup> ones, reducing the amount of **alkali** and introducing **acetic acid** are effective strategies for **reducing discharge intensity**.
- Coating **pore sealing** for increasing the corrosion resistance of the PEO coatings.

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